Roll No.

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M.Sc. (Second Semester) EXAMINATION, May - June, 2022 CHEMISTRY

Paper No. CH - 8 (Reaction Mechanism)

Time: Three Hours] [Maximum Marks: 80

Note: Attempt all the sections as directed.

(Section - A)

(Objective/Multiple Choice Questions)

(1 mark each)

1. Using the following code, arrange the given halides in order to decreasing reactivity towards NaI in acetone:

- I. MeCl
- II. Me₂CHCl III. Me₂CHF
- (A) I > II > III
- (B) II > III > I
- (C) II > I > III
- (D) III>II>I

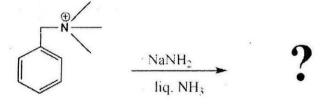
2. Benzyne is generated from:

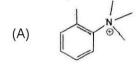
(A) Chlorobenzene in presence of sodamide in liquid NH_3

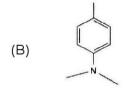
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- (B) Benzene with sodium in liquid NH₃
- (C) Benzene in liquid NH₃
- (D) Action of heat on benzoic acid
- 3. Which of the following statements is incorrect for SN₂ reaction?
 - (A) In the gas phase, the order of reactivity of halide ions is: $F^->C^->B$ $r^->I^-$ while in methanol, the order of reactivity is $F^->B$ $r^->C$ $I^->F$
 - (B) In an aprotic solvent, e.g. DMF, the order of reactivity of halide ion is :F>Cl>Br>I
 - (C) Increase the solvent polarity cause increase in the rate of reaction between OH- and MeS⁺.
 - (D) Increase the solvent polarity casue large increase in the rate of reaction between NH₃ and EtBr.

4. Identify the product of the given reaction?







5. Identify the product of the given reaction?

- 6. Which among the following undergoes nitration most readily?
 - (A) Benzene
 - (B) Acetanilide
 - (C) Acetophenone
 - (D) Chlorobenzene
- 7. Identify the product of the given reaction?

$$\begin{array}{c|c}
 & C_6H_5 \\
 & N-CHO \\
\hline
 & H_2O
\end{array}$$

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- 8. Which one among the following will be least reactive in SE2 (back) reaction (L = leaving group)?
 - (A) Me-L
 - (B) $Me_2CH L$
 - (C) Me_3C-L
 - (D) Et-L
- 9. Which CHCl₃ is more reactive than CHF₃ in SE1 reaction?
 - (A) Cl has larger size than F
 - (B) *C1* uses its d orbital for stabilizing the carbanion formed, where as F can not
 - (C) Cl is less electronegative than F
 - (D) None of these
- 10. Arrange the following compounds in decreasing order of their electrophilic sddition reaction with HX.

(I)
$$C_6H_5$$
- C_6H_2 (II) C_6H_5 C_{H_3} (III) C_6H_5 C_{H_3} (IV) H_2C - C - NO_2

- (A) IV > I > II > III
- (B) III > II > I > IV
- (C) II > III > I > IV
- (D) I > III > IV > I
- (E) None of these

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11. Identify the product of the given reaction?

12. What is the product of the given reaction?

$$C_{6}H_{5} \xrightarrow{O} C_{2}H_{5} + \underbrace{\begin{array}{c}O\\O\\O\\O\\O\end{array}} C_{2}H_{5}ONa \xrightarrow{C_{2}H_{5}ONa}$$

(A)
$$C_6H_5$$
 OC_2H_5 OC_2H_5

(B)
$$C_6H_5$$
 OC_2H_5 $CH(COOC_2H_5)_2$

(C)
$$C_{6}H_{5}$$
 OH $C_{2}H_{5}OOC)_{2}HC$

(D)
$$C_6H_5$$
 OH $CH(COOC_2H_5)_2$

- 13. Which of the following reagents will give syn addition reaction with alkene?
 - (i) Br₂
 - (ii) Dil.KMnO4/OH
 - (iii) $OsO_4/NaSO_3H/HOH$
 - (iv) $H_2/Ni/\Delta$
 - (A) only i
 - (B) ii and iii
 - (C) ii, iii and iv
 - (D) only iv

(D)

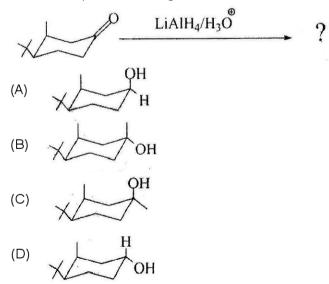
- (E) None of these
- 14. What is the product of the given reaction:

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- 15. Consider the following statements:
 - (1) Alkene is more reactive than alkyne for electrophilic addition
 - (2) Alkyne is more reactive than alkene for Nucleophilic addition
 - (3) Alkyne is more reactive than alkene for electrophilic addition reaction
 - (4) Alkene having CF3 at vinylic carbon is more reactive than alkene having CH3 group of these, correct statement are:
 - (A) 3 and 4
 - (B) 1,2 and 3
 - (C) 1,2 and 4
 - (D) 1 and 2

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16. What is the product of the given reaction:



17. In aldol addition reaction product is always

 β – Hydroxyaldehydes

 β – hydroxyketone

 $\alpha - \beta$ - unsaturated aldehydes

(D) $\alpha - \beta$ - unsaturated ketones

18. The intermediate will form in stobbe condensation is?

19. What is the product of the given reaction is?

$$C_6H_5CHO + < \frac{COOH}{COOH} \qquad \frac{Pyridine}{\triangle}$$



$$(A) \qquad \begin{matrix} H \\ \\ C_0H_5 \end{matrix} \qquad \begin{matrix} COOH \\ \\ COOH \end{matrix}$$

20. Match the following with their respective reagents/catalyst

- Knovenagel (a)
- (i) $POCl_3/H_2O$
- Vilsmeir
- (ii) $NaNH_2/liq.NH_3$
- Gatterman Koach
- (iii) aq.KCN
- Sommlet Hauser
- (iv) Pyridine
- (v) NaOH
- (vi) AlCl₃/CuCl
- a-iv, b-i, c-v, d-ii
- a-iii, b-i, c-vi, d-ii
- a-iv, b-i, c-vi, d-ii
- (D) a-i, b-ii, c-iii, d-iv

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Section - B

(Very Short Answer Type Questions)

(2 marks each)

Note: Attempt all questions.

- 1. Define neighbouring group participation with suitable example.
- 2. Define the regioselectivity.
- 3. How do you identify the reaction is SE1 or SE2?
- 4. Explain the Gottermann Koch reaction.
- Define the enantioselective reaction.
- 6. Explain the chemoselectivity.
- 7. How do you differentiate the properties of LiAlH₄ and NaBH₄?
- 8. What do you mean by cross Aldol condensation reaction?

Section - C

(Short Answer Type Questions)

(3 marks each)

Note: Attempt all questions.

- What is benzyne mechanism explain with reaction mechanism.
- 2. What is Somlet Hauser reaction? Write the Mechanism and suitable example.
- 3. Explain the effect of substrates in aliphatic electrophilic substitution reaction.
- 4. How do you understand the reaction is SE2? Explain with suitable example.
- 5. Explain the addition of water in alkene with suitable example.
- 6. Discuss the addition reaction of cyclopropane ring.
- 7. What is Witting reaction give the mechanism and application?
- 8. What is Knovenegel reaction? Write the mechanism and examples.

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Section - D

(Long Answer Type Questions)

(5 marks each)

Note: Attempt all questions.

What do you mean by ArSN2 and ArSN1 mechanism?
 Explain with suitable example.

OR

What is Smile reaction give the mechanism and application.

2. Explain the SE² and SE¹ Mechanism with suitable example and rate profile.

OR

What is Vilsmier reaction give the mechanism and application.

3. Explain the hydroboration-Oxidation reaction with suitable example.

OR

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Explain the Michael addition reaction with examples and mechanism.

4. What is Benzoin condensation reaction? Write the Mechanism and examples.

OR

What is Stobbe condensation reaction give the mechanism and application.